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Our chemistry experts are ready to help. ... A solution was made by dissolving 12.5 g of Na<sub>3</sub>PO<sub>4</sub> in 125.00 mL of water. The volume of the resulting solution was 128 mL. Calculate the molarity and ...

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easy chemistry help please? A 39.0 mL sample of 1.05 Moles KBr and a 56.0 mL sample of 0.680 Moles KBr are mixed. The solution is then heated to evaporate water until the total volume is 65.0 mL.

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### Science Quiz: Chemistry: Solutions - Ducksters

Chemistry is a big part of your everyday life. You find chemistry in foods, the air, cleaning chemicals, your emotions, and literally every object you can see or touch.. Here are 10 examples of everyday chemistry. Some common chemistry might be obvious, but other examples might surprise you.

### Examples of Chemistry in Everyday Life - ThoughtCo

Place the cabbage leaves in the plastic bottle, half fill the bottle with hot water and screw the lid on tightly. Shake the bottle for a few minutes until the water becomes a deep purple colour. Leave the solution to cool. Strain the solution and add sufficient water to the solution to make about 1 L.

### 8 simple chemistry experiments that your kids can do at ...

Well, you add diluent or solvent to the stock solution in order to obtain a working solution. Say you have a 6 M stock solution of hydrochloric acid (HCl), and you need 100 mL of 0.5 M HCl to work with in the lab. You will need to take some amount of the stock solution and add diluent to it (water, in this case) in order to achieve a less concentrated solution.

### How to Calculate Dilutions | Sciencing

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